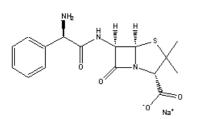
## Catalog Number: 190148, 194199, 194526 Ampicillin, sodium salt

Structure:



Molecular Formula: C<sub>16</sub>H<sub>18</sub>N<sub>3</sub>O<sub>4</sub>SNa

Molecular Weight: 371.4

CAS # 69-52-3

Synonyms: D[-]-a-Aminobenzylpenicillin; 6-[D(-)-a-Aminophenylacetamido]penicillanic acid

Physical Description: White to off-white powder

Ampicillin rapidly loses activity when stored above a pH of 7.0.<sup>4,5,11</sup> Optimal conditions for storage are at +4°C, pH 3.8 to 5. At these conditions the solutions will retain more than 90% activity for approximately one week. Buffer compositions may also effect stability: Optimally Tris buffers should have a pH of approximately 5, citrate buffers at 7, acetate buffers at  $6.^{4,5}$ 

Autoclaving solutions of ampicillin will destroy activity. Solutions can be sterilized by filtration (0.22 um filter) and stored frozen (-20°C or below). Frozen solutions can be stored for up to 3 months.

## pKa Values (at 25°C):12

pKa = 2.5 (-COOH) pKa = 7.3 (-NH<sub>2</sub>)

**Description:** Ampicillin is a semi-synthetic derivative of penicillin, active as a broad-spectrum antibiotic. It is inactivated by betalactamases and for this reason a beta-lactamase inhibitor should be considered when using ampicillin. Against gram-positive bacteria, ampicillin has a similar mode of action as benzylpenicillin; against gram-negative bacteria, it has a similar mode of action as chloramphenicol and tetracyclines. In E. coli it inhibits cell wall synthesis.<sup>18</sup>

Also described for the use of inhancing luminol chemiluminescence.<sup>15</sup>

**Suggested Effective Concentrations:** 100 mg/liter for both gram positive and gram negative bacteria. It is typically stable in media at 37°C for approximately 3 days.<sup>26</sup>

## Availability:

Catalog Number	Description	Size
190148	Ampicillin, sodium salt	5 g 25 g 100g
194199		20 mg 50 mg
194526	Ampicillin, sodium salt, cell culture reagent	5 g 25 g 100 g

**Solubility:** Soluble in water (50 mg/ml); soluble in 1 M Ammonium Hydroxide, dilute acids or bases<sup>12</sup>; practically insoluble in alcohol, chloroform, ether and fixed oils.

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